## WHAT IS CLAIMED IS:

1		1.	A method for removing and recovering an ion from a solution
2	comprising:		
3	_	(a)	contacting said solution with a solid ion extractant comprising a
4	recyclable act	ivated i	on coordinating agent to remove at least a portion of said ion from said
5	solution;		
6		(b)	deactivating said ion coordinating agent;
7		(c)	dissolving said ion in a recovery solvent and removing said ion from
8	said solid ion	extract	ant; and
9		(d)	recovering said ion from said recovery solvent.
1		2.	The method of Claim 1, wherein said ion is an anion.
1		3.	The method of Claim 1, wherein said ion is a perfluoroalkylsulfonate
2	or a perfluoro	oalkyl c	arboxylate.
1		4.	The method of Claim 1, wherein said ion coordinating agent comprise
2	a lipophilic p	ortion.	
1		5.	The method of Claim 1, wherein said active ion coordinating agent is
2	ionizable in a	an aque	ous solution.
1		6.	The method of Claim 1, wherein said step of removing said ion from
2	said solution	compr	ises an ion-exchange process.
1		7.	The method of Claim 1, wherein said ion coordinating agent is an
2	organometal	lic com	pound.
1		8.	The method of Claim 7, wherein said ion coordinating agent is redox-
2	recyclable.	•	
1		9.	The method of Claim 8, wherein said ion coordinating agent is
2	oxidized pri	or to sa	id step (a).
1		10.	The method of Claim 9, wherein said oxidation step comprises an
2	electrochem	ical ox	idation process.

The method of Claim 9, wherein said oxidation step comprises a 11. 1 chemical oxidation process. 2 The method of Claim 8, wherein said deactivation step comprises 12. 1 reducing said ion coordinating agent. 2 The method of Claim 12, wherein said reduction step comprises an 13. 1 electrochemical reduction process. 2 The method of Claim 12, wherein said reduction step comprises a 14. 1 chemical reduction process. 2 The method of Claim 14, wherein said chemical reduction process 15. 1 comprises contacting said solid ion extractant with a deactivating solution. 2 The method of Claim 15, wherein said deactivating solution comprises 16. 1 a compound selected from the group consisting of Na<sub>4</sub>Fe(CN)<sub>6</sub>, K<sub>4</sub>Fe(CN)<sub>6</sub>, Na<sub>2</sub>S<sub>2</sub>O<sub>4</sub>, Cr<sup>2+</sup> 2 salt, V2+ salt and NaBH4. 3 The method of Claim 1, wherein said solid ion extractant is a 17. 1 composition of the formula: 2 z  $Y_a$ 3 wherein 4 each of Ar1 and Ar2 is independently C4-C20 aryl; 5 M is a transition metal; 6

each of X1, X2 and X3 is independently a bond, O, S, or NR4;

each of  $R^2$ ,  $R^3$  and  $R^4$  is independently H, or  $C_1$ - $C_6$  alkyl;

R<sup>1</sup> is C<sub>2</sub>-C<sub>20</sub> alkylene;

Z is a solid support;

Y is an anion; and

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12	a is	s 0 when said ion coordinating agent is deactivated, and
13	a is	s an integer from 1 to 3 when said ion coordinating agent is activated.
1	18.	The method of Claim 17, wherein M is selected from the group
2	consisting of Fe, F	Ru, Mn, Co, Ni, Cr, Os, Rh and Ir.
1	19.	The method of Claim 17, wherein Ar <sup>1</sup> and Ar <sup>2</sup> are selected from the
2	group consisting of	of cyclopentadienyl, dicarbollide and phenyl, each of which can be
3	optionally substitu	ated.
1	20.	The method of Claim 17, wherein each of $X^1$ , $X^2$ and $X^3$ is
2	independently a b	ond or O.
1	21.	The method of Claim 17, wherein $R^2$ and $R^3$ are $C_1$ - $C_6$ alkyl.
1	22.	. The method of Claim 17, wherein said solid support is glass or a
2	polymeric resin.	
1	23.	. The method of Claim 22, wherein said polymeric resin is selected from
2	the group consisti	ng of acrylic ester, polyvinyl, polystyrene, polypyrrole, polyolefin, and
3	polyaromatic.	
1	24	. The method of Claim 17, wherein Y is selected from the group
2	consisting of nitra	ate, halide, HSO <sub>4</sub> , ClO <sub>4</sub> , ReO <sub>4</sub> , PF <sub>6</sub> , carboxylate and CF <sub>3</sub> SO <sub>3</sub> .
1	25	. The method of Claim 17, wherein a is 1 when said ion coordinating
2	agent is activated	
1	26	. The method of Claim 1, wherein said solution is an aqueous solution.
1	27	The method of Claim 26, wherein the temperature of solution of said
2	step (a) is least ab	oout 24 °C.
1	28	. The method of Claim 1, wherein the temperature of said recovery
2.	solution is at least	t about 85 °C

1	29	The method of Claim 28, wherein said recovery step (d) comprises
2	reducing the temp	perature of said recovery solution to less than about 5 °C to precipitate said
3	ion.	
1	30	. The method of Claim 29, wherein said recovery step further comprises
2	separating said pr	recipitated ion from said recovery solution.
1	31	. The method of Claim 1, further comprising:
2	(e	re-activating said deactivated ion coordinating agent; and
3	(f)	repeating said steps (a)-(e).
1	32	2. A method for removing and recovering an anion from an aqueous
2	solution compris	ing:
3	(a	contacting said solution with a solid ion extractant comprising an ion
4	coordinating age	nt to remove at least a portion of said anion from said aqueous solution,
5	wherein said soli	d ion extractant is a composition of the formula:
•		$ \begin{bmatrix}  & & & & & & & & & & & & & & & & & & $
6		
7	wherein	
8		ach of Ar <sup>1</sup> and Ar <sup>2</sup> is independently C <sub>4</sub> -C <sub>20</sub> aryl;
9		I is a transition metal;
10	R	<sup>1</sup> is C <sub>2</sub> -C <sub>20</sub> alkylene;

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that a is 0;

reducing said ion coordinating agent in said solid ion extractant such

each of  $X^1$ ,  $X^2$  and  $X^3$  is independently a bond, O, S, or  $NR^4$ ;

each of  $R^2$ ,  $R^3$  and  $R^4$  is independently H, or  $C_1$ - $C_6$  alkyl;

Z is a solid support;

a is an integer from 1 to 3;

Y is an anion; and

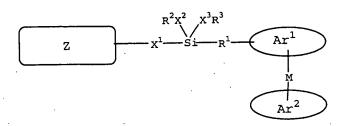
(b)

18	(c)	dissolving said anion in a recovery solvent and removing said anion
19	from said solid ion ex	xtractant; and
20	(d)	recovering said anion.
1	33.	The method of Claim 32, wherein said anion is a perfluoroalkyl
2	sulfonate or perfluor	oalkyl carboxylate.
1	34.	The method of Claim 32, wherein said step of removing said anion
2	from said aqueous so	olution comprises an ion-exchange process.
. 1	35.	The method of Claim 32, wherein a in said step (a) is 1.
1	36.	The method of Claim 32, wherein said ion coordinating agent is redox-
2	recyclable.	
1	37.	The method of Claim 36, further comprising a step of oxidizing said
2	ion coordinating age	ent from a=0 to a=1 prior to said step (a).
1	38.	The method of Claim 37, wherein said oxidizing step comprises an
2	electrochemical oxid	dation process.
1	39.	The method of Claim 37, wherein said oxidizing step comprises a
2	chemical oxidation	process.
1	40.	The method of Claim 36, wherein said deactivation step comprises
2	reducing said ion co	pordinating agent.
1	41.	The method of Claim 40, wherein said reducing step comprises an
2	electrochemical red	luction process.
1	42.	The method of Claim 40, wherein said reducing step comprises a
2	chemical reduction	process.
1	43.	The method of Claim 42, wherein said chemical reduction process
2	comprises contacting	ng said solid ion extractant with a deactivating solution.
1	44.	The method of Claim 43, wherein said deactivating solution comprises
2	a compound selecte	ed from the group consisting of Na <sub>4</sub> Fe(CN) <sub>6</sub> , K <sub>4</sub> Fe(CN) <sub>6</sub> , Na <sub>2</sub> S <sub>2</sub> O <sub>4</sub> , Cr <sup>2+</sup>
3	salt, V <sup>2+</sup> salt and N	

1	45. The method of Claim 32, wherein M is selected from the group
2	consisting of Fe, Ru, Mn, Co, Ni, Cr, Os, Rh and Ir.
1	46. The method of Claim 45, wherein Ar <sup>1</sup> and Ar <sup>2</sup> are selected from the
2	group consisting of cyclopentadienyl, dicarbollide and phenyl, each of which can be
3	optionally substituted.
1	47. The method of Claim 46, wherein each of $X^1$ , $X^2$ and $X^3$ is
2	independently a bond or O.
1	48. The method of Claim 47, wherein $R^2$ and $R^3$ are $C_1$ - $C_6$ alkyl.
1	49. The method of Claim 48, wherein said solid support is glass or a
2	polymeric resin.
1	50. The method of Claim 49, wherein said polymeric resin is selected from
2	the group consisting of acrylic ester, polyvinyl, polystyrene, polypyrrole, polyolefin, and
3	polyaromatic.
1	51. The method of Claim 50, wherein Y is selected from the group
2	consisting of nitrate, halide, HSO <sub>4</sub> , ClO <sub>4</sub> , ReO <sub>4</sub> , PF <sub>6</sub> , carboxylate and CF <sub>3</sub> SO <sub>3</sub> .
1	52. The method of Claim 32, wherein the temperature of said aqueous
2	solution of said step (a) is least about 24 °C.
1	53. The method of Claim 52, wherein the temperature of said recovery
2	solution is at least about 85 °C.
1	54. The method of Claim 53, wherein said recovery step (d) comprises
2	reducing the temperature of said recovery solution to less than about 5 °C to precipitate said
3	anion.
1	55. The method of Claim 54, wherein said recovery step further comprise
2	separating said precipitated ion from said recovery solution.
1	56. The method of Claim 32, further comprising:
2	(e) re-activating said deactivated ion coordinating agent: and

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- (f) repeating said steps (a)-(e).
- 1 57. A process for producing a solid ion extractant composition of the
- 2 formula:



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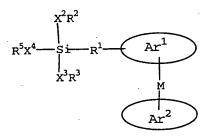
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4 said process comprising contacting the solid support of the composition:



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6 with an ion coordinating agent of the formula:



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8 wherein

9 each of  $Ar^1$  and  $Ar^2$  is independently  $C_4$ - $C_{20}$  aryl;

each of  $X^1$ ,  $X^2$ ,  $X^3$  and  $X^4$  is independently a bond, O, S, or NR<sup>4</sup>;

11  $R^1$  is  $C_2$ - $C_{20}$  alkylene;

each of R<sup>2</sup>, R<sup>3</sup>, R<sup>4</sup> and R<sup>5</sup> is independently H, or C<sub>1</sub>-C<sub>6</sub> alkyl;

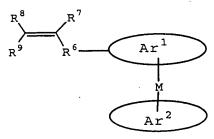
M is a transition metal; and

Z is a solid support.

1 58. The process of Claim 57, further comprising the step of producing said

ion coordinating agent, wherein said ion coordinating agent producing step comprises

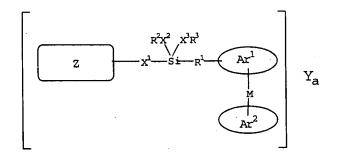
3 contacting an olefin of the formula:



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5	with a silyl compound of the formula $HSi(X^2R^2)(X^3R^3)(X^4R^5)$ in the presence of a catalyst to		
6	produce said ion coordinating agent,		
7	wherein		
8	R <sup>6</sup> is a bond or C <sub>1</sub> -C <sub>18</sub> alkylene; and		
9	each of $R^7$ , $R^8$ and $R^9$ is independently H or $C_1$ - $C_6$ alkyl.		
1	59. The method of Claim 58, wherein said catalyst is selected from the		
2	group consisting of Karstedt's catalyst, Speier's catalyst, other silylplatinum complexes,		
3	transition metal nanoclusters, dicobalt octacarbonyl, nickel tetracarbonyl, Wilkonson's		
4	catalyst, chromium hexacarbonyl, and zirconocenes.		
1	60. The process of Claim 57, wherein M is selected from the group		
2	consisting of Fe, Ru, Mn, Co, Ni, Cr, Os, Rh and Ir.		
1	61. The process of Claim 60, wherein Ar <sup>1</sup> and Ar <sup>2</sup> are selected from the		
2	group consisting of cyclopentadienyl, dicarbollide and phenyl, each of which can be		
3	optionally substituted.		
1	62. The process of Claim 61, wherein each of $X^1$ , $X^2$ and $X^3$ are		
2	independently a bond or O.		
1	63. The process of Claim 62, wherein $X^4$ is O.		
· 1	64. The process of Claim 63, wherein $R^2$ , $R^3$ and $R^5$ are $C_1$ - $C_6$ alkyl.		
1	65. The process of Claim 64, wherein said solid support is glass or a		
2	polymeric resin.		
1	66. The process of Claim 65, wherein said polymeric resin is selected from		
2	the group consisting of acrylic ester, polyvinyl, polystyrene, polypyrrole, polyolefin, and		
3	polyaromatic.		
1	67. The process of Claim 57, further comprising oxidizing said solid ion		
2	extractant to produce an activated solid ion extractant of the formula:		



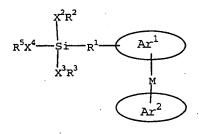
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4 wherein

5 Y is an anion; and

6 a is an integer from 1 to 3.

- 1 68. The process of Claim 67, wherein a is 1.
- 1 69. The process of Claim 67, wherein Y is selected from the group consisting of nitrate, halide, HSO<sub>4</sub>, ClO<sub>4</sub>, ReO<sub>4</sub>, PF<sub>6</sub>, carboxylate and CF<sub>3</sub>SO<sub>3</sub>.
- The process of Claim 67, wherein said oxidizing step comprises an electrochemical oxidation process.
- The process of Claim 67, wherein said oxidizing step comprises a chemical oxidation process.
- The process of Claim 71, wherein said chemical oxidation process comprises contacting said solid ion extractant with a solution comprising an oxidant to
- 3 produce said activated solid ion extractant.
- The process of Claim 72, wherein said oxidant is selected from the
- group consisting of  $Fe(NO_3)_3$ ,  $AgNO_3$ ,  $FeCl_3$ , AgF, NaOCl,  $Ce(NH_4)_2(NO_3)_6$  and  $Ce(SO_4)_2$ .
  - 74. A compound of the formula:



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3 wherein

4 each of  $Ar^1$  and  $Ar^2$  is independently  $C_4$ - $C_{20}$  aryl;

5 each of  $X^1$ ,  $X^2$ ,  $X^3$  and  $X^4$  is independently a bond, O, S, or  $NR^4$ ;

5	R <sup>1</sup> is C <sub>2</sub> -C <sub>20</sub> alkylene;
7	each of $\mathbb{R}^2$ , $\mathbb{R}^3$ , $\mathbb{R}^4$ and $\mathbb{R}^5$ is independently H, or $\mathbb{C}_1$ - $\mathbb{C}_6$ alkyl; and
8	M is a transition metal.
1	75. The compound of Claim 74, wherein M is selected from the group
2	consisting of Fe, Ru, Mn, Co, Ni, Cr, Os, Rh and Ir.
1	76. The compound of Claim 74, wherein Ar <sup>1</sup> and Ar <sup>2</sup> are selected from the
2	group consisting of cyclopentadienyl, dicarbollide and phenyl, each of which can be
3	optionally substituted.
1	77. The compound of Claim 74, wherein each of $X^1$ , $X^2$ , $X^3$ and $X^4$ is
2	independently a bond or O.
1	78. The compound of Claim 74, wherein R <sup>2</sup> and R <sup>3</sup> are C <sub>1</sub> -C <sub>6</sub> alkyl.

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